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Answers

3. Write the Lewis structures. The C-O bond order in methanol is one. The C-O bond order in formaldehyde is two. The C-O bond order in the carbonate ion is $4/3$. Therefore, the C-O bond length in the carbonate ion should be closer to 143

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pm than to 116 pm.

Chem;GI;Answers

CA 3 Practice Problem
Solutions ChemActivity
3 Exercises 1-3 1. 5.47
10 -18 J. 2. a) $IE_a = \frac{-2(-1)}{d_1} = \frac{2}{d_1}$ b)
 $IE_b = \frac{-1(-1)}{2d_1} = \frac{1}{2d_1}$ $IE_a > IE_b$ 3.
The ionization energy
of case (a) is larger,
1.20 k / d₁, than that
of case (b), 1.17 k / d₁.

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**Practice - 5th ed -
CA 3 Practice
Problem ...**

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3, 4, 5, 8, 9, 10, 11:

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**Homework AP - Mr.
Apthorp Chemistry**

ChemActivity 3 1 547

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$\times 10^{-18}$ J Several possible answers A typical one would have all three electrons at a much farther distance than in H from the nucleus - at least six or seven times as far

Problem 1 a) $V = b)$

The IE of He is slightly less than twice the IE of H because the electron-electron

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Answers to
Page 7/26

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ChemActivity 1 - The Nuclear Atom- Chapter 5

1. 6 2. 6,7,7 3. 6,6,7

4. (a) neutral same e and p, ion different e and p (b) assign +1 to each proton and -1 to each electron and take the difference.

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Answers

3. The calculated mass of ^{12}C based on the masses of the constituent particles is 12.099 amu; and the

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actual mass of ^{12}C is exactly 12 amu.

ChemActivity 3 1. 5.47

$\times 10^{-18}$ J. 2. a) $IE_a =$

$-(2)(-1)/d_1 = 2/d_1$ b)

$IE_b = -(1)(-1)/2d_1 =$

$1/2d_1$ $IE_a > IE_b$ 3. The

ionization energy of

case (a) is larger, 1.20

kJ/d , than that of case

(b), 1.17 kJ/d .

Problems

Answers to Exercises and Problems

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Coulombic Potential
Energy. STUDY.
Flashcards. Learn.
Write. Spell. Test.
PLAY. Match. Gravity.
Created by.
kasha_pasha.
Introduction to
Potential Energy and
Ionization Energy.
Terms in this set (11)
What is potential
energy (V)?

**Chemactivity 3:
Coulombic Potential
Energy Flashcards ...**

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View Test Prep - Chem Activity 3 from CHEM 1330 at University of Arkansas, Pine Bluff. ChemActivity 3 Atoms and the Periodic Table (What are atoms?) Model 1: Schematic diagrams for various atoms 0

Chem Activity 3 - ChemActivity 3 Atoms and the Periodic ...

ChemActivity 39

ChemActivity 40

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ChemActivity 40a

ChemActivity 40b. F,
Sep 9 . Details of the
equilibrium constant .

CA 40c: CTQ a-i .

ChemActivity 40c. M,
Sep 12 . Solution
Concentration .

ChemActivity Solution
Concentrations: CTQ

1-10 . ChemActivity

S.C. W, Sep 14 .

Equilibrium of insoluble
salts, ChemActivity 41:

CTQ 1-13, Ex. 6 . CA ...

Chem 124

Page 12/26

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**Homework - Wofford
College**

3. What is the mass (in grams) of 4.35×10^6 atoms of ^{12}C ? 4. What is the mass (in grams) of 6.022×10^{23} atoms of ^{12}C ? 5. What is the mass (in grams) of one molecule of methane which has one ^{12}C atom and four ^1H atoms, $^{12}\text{C}^1\text{H}_4$? 6. a) Define mass number. b) Define atomic number. 4

ChemActivity 1 The
Page 13/26

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Nuclear Atom

The Nuclear Atom

ChemActivity 3

Coulombic Potential

Energy 9 5. Recall that a 1H atom consists of a proton as the nucleus and an electron outside of the nucleus. Is the potential energy, V , of a hydrogen atom a positive or negative number? Explain your answer. Model 2: Ionization Energy.

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ANSWERS TO CHEMACTIVITY 32 MOLARITY PDF

(See chemactivity 5 for explanation) Two atoms have the same nuclear charge, but one atom has a lower IE than the other atom. How could this happen? The first atom must have a greater distance between the valence electron and the nucleus than the other atom. **different

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distances change
results.

Chemactivity 4: The Shell Model (I)

Flashcards | Quizlet

Color Printable Periodic
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everything you need
that can fit on a page
and still be readable.
Color table with atomic
numbers, element
symbols, element
names, atomic
weights, periods, and
groups. [2013 Edition]

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[2012

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- Black/white table with

atomic numbers,

element symbols,

element names, atomic

weights, periods.

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The answers to the

questions tend to be

sequential - you may

need the answer to

question 3 to answer

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question 4, for example. d.) For each question, the group must decide on an answer that will be submitted for the group. ... After you have finished (and submitted) the ChemActivity, the answers to the questions will be available in the ...

ChemActivity

General Directions

All of the answers can

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be found online for almost every single POGIL..... Currently, there is a significant amount of discussion on teaching list serves about the frustration of people posting answer keys online and students checking the internet instead of doing the work. Many teachers, myself included, experience this frustration.

POGIL, Posting Keys

Page 20/26

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Online, Cheating and Checkpoints ...

ChemActivity 46:

ChemActivity 46: Oct

3: Buffer Solutions: CA

- Buffers(a) CA -

Buffers (b) CA -

Buffers: Oct 5: Exam 2:

Acid Base Equilibrium :

Oct 6: Titrations: CA -

Titration: CA -

Titration: Oct 7:

Polyprotic systems: CA

- Polyprotic Acids: Oct

10: Introduction to

Thermodynamics and

the 2nd Law:

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Chem 124
ChemActivity
Homework - Wofford
College

March 3, 2014 This homework is practice on formulas and naming of ionic compounds.

Activity3_Homework
February 19, 2014 This is the Chem Talk and practice problems (homework) for Chapter 6 Activity 1 comparing the

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characteristics of physical and chemical changes.

Activity1_ChemTalk_HW February 10, 2014

Active Chemistry Assignments - Ms. Steensen's Science Classes

the data in Table I of ChemActivity 10: Electron Configurations. Model 2. Simulated Photoelectron Spectrum of Scandium.

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(The 1s peak occurs at 433 MJ/mole and is not shown in this spectrum.) 48.5 39.2
Ionization Energy (MJ/mole) 5.44 3.24
0.63 0.77 Critical Thinking Question 3.

Electron Configurations and the PT answer key

Aluminum adopts a $+3$ charge in an ionic compound, so the formula for aluminum nitrate is

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$\text{Al}(\text{NO}_3)_3$.

The balanced equation is: ... Answer a The products are $\text{ZnCl}_2 + \text{Fe}$ Answer b No reaction. Summary. Metals and halogens are ranked according to their ability to displace other metals or halogens below them in the series.

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ecf8427e.